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Thermodynamic boundary conditions in molecular simulation

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Definition of a model for a computer simulation

For any system four choices have to be made



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Molecular simulation: algorithms, boundary conditions and constraints

1. Generating configurational ensembles

a. Integrating equations of motion

- 1. Newton: molecular dynamics simulation (MD)
- 2. Langevin: stochastic dynamics simulation (SD)

b. Monte Carlo type approach

- 1. Metropolis Monte Carlo simulation (MC)
- 2. Replica exchange technique (RE)

2. Boundary conditions

- 1. Spatial boundary conditions
- 2. Thermodynamic boundary conditions
 - temperature
 - pressure
 - other

3. Constraints

- 1. Distance constraints
- 2. Other constraints

4. MD algorithm

Molecular Dynamics with Coupling to an External Bath

Standard (micro-canonical) MD: independent parameters:

number of particles, N volume, V total energy, E_{tot}

conserved quantities:

 E_{tot} \bar{P}_{tot} = total momentum L_{tot} = total angular momentum (only in vacuo, **not** with PBC)

dependent quantities:

temperature = Tpressure = P

Why apply MD at constant T and/or P?

-study system properties as a function of T and P rather than E and V(normally one measures at constant T and P) -long-range Coulomb force: ~ r^{-2} noise \rightarrow rising T use of cut-off radius introduces heat into the system:

positions

-study of non-equilibrium systems: maintain a P or T gradient.

by scaling the *velocities* **Couple:** *temperature* by scaling the pressure

Definition of temperature and pressure

• Macroscopic temperature, T, is defined from the average kinetic energy per degree of freedom

$$\mathcal{K}(N,\mathbf{p},\mathbf{q}) = 3N \frac{k_{B}T}{2}$$

• From this we define an instantaneous temperature

$$\mathcal{T}(N,\mathbf{p},\mathbf{q}) = \frac{2\mathcal{K}(N,\mathbf{p},\mathbf{q})}{3Nk_B}$$
 with $T = \langle \mathcal{T}(N,\mathbf{p},\mathbf{q}) \rangle$

• Virial theorem

$$\mathbf{K} = W_{tot} = W + \frac{3}{2}PV$$

• In terms of instantaneous properties

$$\mathcal{W}(N,\mathbf{q}) = -\frac{1}{2} \sum_{i=1}^{N} \mathbf{r}_{i} \otimes \mathbf{F}_{i}(\mathbf{r})$$
$$\mathcal{P}(N,\mathbf{q},\mathbf{p}) = \frac{2\left[\mathcal{K}(N,\mathbf{q}) - \mathcal{W}(N,\mathbf{q})\right]}{3\mathcal{V}(N,\mathbf{q})}$$

 $\mathcal{W}(N,\mathbf{q}) \text{ and } \mathcal{P}(N,\mathbf{q},\mathbf{p}) \text{ are tensors}$ $\mathcal{K}(N,\mathbf{p}) \text{ is written as diagonal}$ $\mathcal{K}(N,\mathbf{p}) = \sum (2m_i)^{-1} \mathbf{p}_i \otimes \mathbf{p}_i$ with $P = \langle \operatorname{Tr}(\mathcal{P}(N,\mathbf{q},\mathbf{p})) \rangle$

Trace: the sum of the diagonal elements

Thermostats

Four ways to control the temperature T(t)

- 1. Constraint method (isokinetic simulation) $T(t) = T_{ref}$
- 2. Stochastic method (Langevin dynamics)

 $m_{i}\frac{d\vec{v}_{i}(t)}{dt} = \vec{f}_{i}(t) - m_{i}\gamma_{i}\vec{v}_{i}(t) + \vec{f}_{i}^{st}(t) \qquad \left\langle \vec{f}_{i}^{st}(0)\vec{f}_{j}^{st}(t) \right\rangle = 6m_{i}k_{B}T_{ref}\gamma_{i}\delta_{ij}\delta(t)$

Distorts the dynamics of individual particles

3. Extended system method (Nosé-Hoover)

$$m_{i} \frac{d\vec{v}_{i}(t)}{dt} = \vec{f}_{i}(t) - m_{i} \gamma_{NH} \vec{v}_{i}(t) \qquad \qquad \frac{d\gamma_{NH}(t)}{dt} = (\tau_{T,NH})^{-2} (\frac{T(t)}{T_{ref}} - 1)$$

May introduce spurious oscillations

4. Weak-coupling method

$$m_{i}\frac{d\vec{v}_{i}(t)}{dt} = \vec{f}_{i}(t) + m_{i}(2\tau_{T})^{-1}(\frac{T_{ref}}{T(t)} - 1)\vec{v}_{i}(t)$$

$$\frac{dT(t)}{dt} = \tau_T^{-1} \Big[T_{ref} - T(t) \Big]$$

Statistical-mechanical ensemble ?

Molecular dynamics with (weak) coupling to a temperature bath

Variety of methods:

- 1. Constraint method
- 2. Stochastic method Langevin equation
- 3. Extended system method
- 4. Weak coupling method

- Hamiltonian, but non-physical
- sizeable effects on atomic trajectories

2nd order coupling, spurious oscillations

$$\frac{dT(t)}{dt} = \frac{1}{\tau_T} \left[T_{bath} - T(t) \right]$$

$$\Delta E_{kin} = (\lambda^2(t) - 1) \frac{1}{2} N_{df} k_B T(t)$$

$$= N_{df} c_V^{df} \Delta T$$

$$\lambda(t) = \left[1 + \frac{2c_V^{df}}{k_B} \frac{\Delta t}{\tau_T} \left[\frac{T_{bath}}{T(t)} - 1 \right] \right]^{\frac{1}{2}}$$

 $\frac{dE_{kin}(t)}{dt} = \text{constant}\left[T_{bath} - T(t)\right]$

first order, physical, but ensemble ?

Energy fluctuations in liquid water as function of the coupling strength τ_{T}



H.J.C. Berendsen et al., J. Chem. Phys., 81 (1984) 3684 - 3690

Coupling to a heat bath at $T = T_{ref}$: stochastic methods

1. Collision model:

The velocity of a randomly selected particle after each time period $\tau_{collision}$ is chosen afresh from a Maxwell-Boltzmann velocity distribution at $T = T_{ref}$

This results in:

$$\langle T \rangle = T_{ref}$$

2. Langevin model:

A frictional force $-m_i \gamma_i \vec{v}_i$ and a stochastic force \vec{f}_i^{st} are added to the particle equations of motion:

$$m_i \frac{d\vec{v}_i(t)}{dt} = \vec{f}_i(t) - m_i \gamma_i \vec{v}_i(t) + \vec{f}_i^{st}(t)$$

$$/(\vec{c}_{st})^2$$

This results in:

$$\langle T \rangle = \frac{\left\langle \left(\vec{f}_i^{st} \right)^2 \right\rangle}{6m_i \gamma_i k_B} = T_{ref}$$

Energy distributions for different thermostats

Liquid argon (1000 atoms), periodic boundary conditions, 80 K, 1400 kgm⁻³



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large τ_T : *micro-canonical* distribution (width E_{not} < canonical width) **NH** and **LD**: *canonical* distribution independent of τ_{NH} or γ

Atom velocity auto-correlation function for different thermostats

Liquid argon (1000 atoms), periodic boundary conditions, 80 K, 1400 kgm⁻³



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WC and NH: no distortion of atom dynamics (global *T*-coupling) **Langevin Dynamics**: *distortion of dynamics increases with* γ

Potential energy auto-correlation function for different thermostats

Liquid argon (1000 atoms), periodic boundary conditions, 80 K, 1400 kgm⁻³



Z. Lin & W.F. van Gunsteren, J. Chem. Phys. 143 (**2015**) 034110

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WC: no correlation irrespective the coupling strength (1st order coupling) **NH:** spurious oscillations depending on coupling strength (2nd order coupling) **LD:** correlation increases with size of the friction γ

Issues with thermostats

- Temperature is a system property
 - We don't know the temperature of individual particles

Hot solvent, cold protein

- Solvent consists of many degrees of freedom

$$T_{prot} = 200 \text{ K}$$

$$T_{solv} = 301 \text{ K}$$

$$T_{tot} = 298 \text{ K}$$

$$T_{tot} = 298 \text{ K}$$

$$T_{tot} = \left[N_{dof}^{prot} \mathcal{T}_{prot} + N_{dof}^{solv} \mathcal{T}_{solv} \right] / N_{dof}^{tot}$$

- Due to rotational freedom, the solvent molecules are more prone to cut-off noise than the solute molecule
- This leads to stronger heating up of the solvent
- This heat does not leak away to the solute (insufficient solute solvent collisions)

Remedy

Separately couple the protein and the solvent to individual heat baths

Issues with thermostat

• Flying ice-cube problem

- Kinetic energy of uncoupled degrees of freedom accumulates
 - No interconversion of energy through collisions
- Kinetic energy accumulates in translational degrees of freedom and depletes from internal degrees of freedom



similarly in vacuum simulations: the protein turns into a rotating rod

Possible solutions

- do not apply scaling to (molecular) translational degrees of freedom
- couple translational and internal-rotational degrees of freedom separately
- remove centre-of-mass motion from time to time (discontinuity)
- apply roto-translational constraints to the protein

Molecular Dynamics with coupling to a pressure bath

Variety of methods:

- 1. Constraint method
- 2. Extended system method

Hamiltonian, but not physical

2nd order coupling spurious oscillations

3. Stochastic method

 $M_{V}\ddot{V}(t) = -\left[P_{bath} - P(t)\right] - M_{V}\gamma_{V}\dot{V}(t) + f_{V}^{stoch}(t)$

with

$$\left\langle f_{V}^{st}(0)f_{V}^{st}(t)
ight
angle =2M_{V}k_{B}T_{bath}\gamma_{V}\delta(t)$$

4. Weak coupling method

$$\frac{dP(t)}{dt} = \frac{1}{\tau_{P}} \left[P_{bath} - P(t) \right]$$
$$\Delta P = -\frac{1}{\beta_{P}V} \Delta V$$

1st order, physical

1st order, physical

$$\mu(t) = \left[1 - \frac{\beta_T \Delta t}{\tau_P} \left[P_{bath} - P(t)\right]\right]^{\frac{1}{3}}$$

Virial theorem:

$$PV = \frac{2}{3} \left[\left\langle E_{kin} \right\rangle - \left\langle W \right\rangle \right] =$$

$$PV = \frac{2}{3} \left[\left\langle \frac{1}{2} \sum_{i=1}^{N} m_i \vec{v}_i^2 \right\rangle - \left\langle -\frac{1}{2} \sum_{i

$$\vec{r}_{ij} = \vec{r}_i - \vec{r}_j \qquad \vec{f}_{ij} = \text{force on atom } i \text{ by atom } j$$
Derivations:$$

1. $P = -\frac{\partial A}{\partial V}\Big|_{T} = k_{B}T \frac{\partial \ln Z}{\partial V}\Big|_{T}$, i.e. from the partition function

2. via Newton: $m_i \dot{v}_i = f_i$

with f from **external** versus internal forces

3. via **momentum flux** through a surface

W.G. Hoover, Computational Statistical Mechanics (Elsevier, Amsterdam, 1991)

Unit cell dimensions as a function of temperature

Crystals of a cytidine-derivative

(3',5'-o-(tetra isopropyl-1,3-disiloxanediyl)-cytidine)

MD at constant temperature and pressure

Dimension of box (nm)	T ₀ =113K		T ₀ =289K	
	exp.	MD (50ps)	exp.	MD (160ps)
2a	1.7884	1.792	1.7776	1.790
2b	2.0648	2.050	2.0774	2.064
С	2.7576	2.772	2.8645	2.822
Volume (nm ³)	10.183	10.180	10.563	10.430

Temperature effect qualitatively reproduced

Atomic van der Waals radii are on average correct (GROMOS)

MD at constant temperature, pressure

Temperature

• If there is a small flow of kinetic energy between weakly coupled parts of the system (equipartition not maintained using weak coupling) then

couple subsystems separately to T-baths

Pressure

- The pressure fluctuates much more than the temperature. difference between 2 very large numbers kinetic energy (expansion) and virial (attractive forces)
- Isotropic coupling (uniform scaling in x,y,z directions)
- Anisotropic coupling (e.g. for membrane simulations)
- Pressure depends on temperature:

P(t) = 2/3 [Ekin(t) - W(t)] / V(t)

choose $\tau_{pressure} > \tau_{temperature}$ for couplings to P- and T-baths, otherwise spurious oscillations may occur

Weak-coupling barostat



- The pressure can be given as a tensor, then also µ, the spatial scaling factor, is a tensor
- scaling is a matrix multiplication of the vector **r** and the box vector
- isotropic pressure scaling: same pressure in all directions
- anisotropic pressure scaling: different pressure in x, y and z
- fully anisotropic scaling: the box shape can change
- combinations of the above: x and y coupled to one pressure, (semi-isotropic) z coupled separately (membranes)

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MD leap-frog algorithm with: periodicity constraints ≻all **T-scaling**

- P-scaling **1. Evaluate** $f_i(t)$ from $V(\{x_i(t)\})$ (periodicity) (put particles in box) (calculate P(t) from virial)
- **2. Compute velocities:** $v_i(t + \Delta t/2) = v_i(t \Delta t/2) + m_i^{-1}f_i(t)\Delta t$
- 3. Scale velocities:

$$v_i \left(t + \Delta t/2 \right) = v_i \left(t + \Delta t/2 \right) \cdot \left[1 + \frac{2c_v^{df}}{k_B} \frac{\Delta t}{\tau_T} \left\{ \frac{T_{ref}}{T(t - \Delta t/2)} - 1 \right\} \right]^{\frac{1}{2}}$$

- $x_i(t + \Delta t) = x_i(t) + v_i(t + \Delta t/2)\Delta t$ 4. Compute positions:
- 5. Satisfy constraints: $({x_i(t)}, {x_i^{uc}(t + \Delta t)}, {x_i(t + \Delta t)})$ (periodicity) SHAKE **constrained velocities:** $v_i(t + \Delta t/2) = \left[x_i(t + \Delta t) - x_i(t) \right] / \Delta t$
- 6. Compute temperature: $T(t + \Delta t/2) = \sum_{i=1}^{3N} \frac{1}{2} m_i v_i^2 (t + \Delta t/2) / (N_{df} \frac{1}{2} k_B)$
- 7. Scale positions: scale box lenghts: a, b, c $x_{i}(t + \Delta t) = x_{i}(t + \Delta t) \cdot \left[1 - \frac{\beta_{T} \Delta t}{\tau_{D}} \left[P_{0} - P(t)\right]\right]^{\frac{1}{3}}$

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Spatial distribution of licences GROMOS biomolecular simulation software



GROMOS = Groningen Molecular Simulation + GROMOS Force Field

Generally available: <u>http://www.gromos.net</u>